

Empirical Formula and Molecular Formula Introduction

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{youtube}wnRaBWvhYKY{/youtube}

In this video explains what empirical formula and molecular formula are, how they are different, and we'll learn how to write the empirical formula for a compound when you are given the molecular formula. Molecular formulas tell you how many atoms of each element are in a compound, and empirical formulas tell you the simplest or most reduced ratio of elements in a compound. If a compound's molecular formula cannot be reduced any more, then the empirical formula is the same as the molecular formula. Also, many compounds with different molecular formula have the same empirical formula.

□□□□ : [Tyler DeWitt](#)

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