Pseudo first order reaction

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{youtube}nCvv9x3CJ7s?{/youtube}

In this video explains Pseudo first order reaction:
A reaction which experimentally follows first order but from theoretical rate equation its order is more than one is called pseudo first order reaction Example: 1 C_2H_5I + H_2O → C_2H_5OH + HI Rate =K* C_2H_5I Concentration of water is taken unity i.e it does not get altered during the reaction. So, experimental or actual rate is as follows Rate = K *C_2H_5I

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